

Molality Practice

1. Calculate the molarity of each of the following solutions:

a. 12.4 g KCl in 289.2 mL solution

$$M = \frac{n}{V}$$
$$12.4 \text{ g KCl} \times \frac{1 \text{ mole KCl}}{74.6 \text{ g KCl}} = 0.166 \text{ mole KCl}$$

$$289.2 \text{ mL} \times \frac{1 \text{ L}}{1000 \text{ mL}} = 0.2892 \text{ L}$$

$$\frac{0.166 \text{ mole}}{0.2892 \text{ L}} = 0.57 \text{ M}$$

Practice Problem

b. 16.4 g CaCl_2 in 0.614 L solution

$$M = \frac{n}{V}$$
$$16.4 \text{ g CaCl}_2 \times \frac{1 \text{ mole CaCl}_2}{111.1 \text{ g CaCl}_2} = 0.148 \text{ mole CaCl}_2$$

$$M = \frac{0.148 \text{ mole}}{0.614 \text{ L}} = 0.24 \text{ M}$$

Molarity Molality Practice Problems Answers

David W. Oxtoby, Norman H. Nachtrieb

Molarity Molality Practice Problems Answers:

Whispering the Techniques of Language: An Psychological Quest through **Molarity Molality Practice Problems Answers**

In a digitally-driven earth wherever monitors reign supreme and instant conversation drowns out the subtleties of language, the profound strategies and psychological nuances concealed within words usually go unheard. Yet, located within the pages of **Molarity Molality Practice Problems Answers** a captivating literary prize sporting with organic thoughts, lies a fantastic quest waiting to be undertaken. Composed by a talented wordsmith, this enchanting opus attracts viewers on an introspective trip, lightly unraveling the veiled truths and profound affect resonating within the material of every word. Within the mental depths of the poignant review, we shall embark upon a genuine exploration of the book is key themes, dissect its interesting publishing design, and fail to the effective resonance it evokes serious within the recesses of readers hearts.

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