

Name: _____

Lab Instructor: _____

PREPARATION FOR CHEMISTRY LAB: MEASUREMENT (Part I)

Pre-lab questions need to be completed prior to your coming to lab. They may be collected at the beginning of the laboratory period.

1. What are the units of mass, length, and temperature in the International System of Units (Table 1.2 in your textbook)?
2. Perform the following calculation and report the answer to the correct number of significant figures (assume all quantities are measured):
 $4.11 + 8.9 + 0.1060$
3. Assuming all numbers come from measurements, perform the following calculation and report the answer to the correct number of significant figures
 $2.415 \times 8.6 \times (2.08 \times 10^5)$
4. Calculate the number of mm that are in 538.3 inches. Use $2.54 \text{ cm} = 1 \text{ in}$ and show your work.
5. If you have 3.6 cm^3 of water, how many liters of water do you have?

Preparation For Chemistry Lab Measurement Part I Number

C Cleary



Preparation For Chemistry Lab Measurement Part I Number:

Whispering the Secrets of Language: An Mental Quest through **Preparation For Chemistry Lab Measurement Part I Number**

In a digitally-driven earth wherever monitors reign great and immediate connection drowns out the subtleties of language, the profound techniques and psychological subtleties concealed within phrases usually move unheard. Yet, located within the pages of **Preparation For Chemistry Lab Measurement Part I Number** a fascinating literary treasure pulsating with organic thoughts, lies an exceptional journey waiting to be undertaken. Written by an experienced wordsmith, this marvelous opus invites readers on an introspective journey, gently unraveling the veiled truths and profound influence resonating within ab muscles cloth of every word. Within the mental depths with this poignant evaluation, we shall embark upon a genuine exploration of the book is key themes, dissect its interesting publishing model, and yield to the powerful resonance it evokes deep within the recesses of readers hearts.

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