



- a. A
b. B

- c. C
d. D

- ___ 10. The energy as heat produced by a reactor is used to
- boil water for steam turbines.
 - melt metal.
 - produce graphite.
 - produce coal.
- ___ 11. At present, fusion reactions
- cannot be used to produce energy in reactors.
 - produce the energy in some nuclear power plants.
 - produce the energy in most nuclear power plants.
 - produce the energy in all recent nuclear power plants.
- ___ 12. What is the product of β -ray emission from a radioactive isotope of lead?
- Thallium
 - Mercury
 - Bismuth
 - Lead
- ___ 13. What is the reason for the decay of naturally radioactive elements?
- To reduce the number of neutrons as the elements lie above the band of stability.
 - To reduce the number of protons as the elements lie below the band of stability.
 - To reduce the number of either neutrons or protons to lie in the band of stability.
 - To release the energy from nucleus in the form of gamma radiation.
- ___ 14. How does a positron differ from an electron?
- A positron has no electrical charge, unlike an electron.
 - A positron has a charge opposite that of an electron.
 - A positron has less mass than an electron.
 - A positron has greater mass than an electron.
- ___ 15. What is the main product of the fusion reactions in our sun?

Prentice Hall Section 25 Nuclear Chemistry Answers

Daniela Niemeyer



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Prentice Hall Section 25 Nuclear Chemistry Answers Book Review: Unveiling the Magic of Language

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