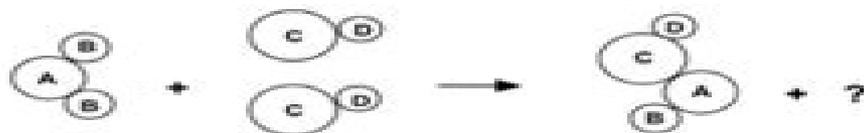


## Chemistry – Unit 7 Review: Chemical Reactions

- \_\_\_ 1. Numbers placed in front of formulas in an equation are called
- \_\_\_ 2. New substances formed in a chemical reaction are called
- \_\_\_ 3. The total number of atoms represented by the formula  $\text{Fe}_3(\text{PO}_4)_2$  is
- \_\_\_ 4. The symbol that indicates a substance dissolved in water is
- \_\_\_ 5. Two white powders are mixed in a flask. The bottom of the flask feels cooler. This an example of what kind of reaction?
- \_\_\_ 6. The charge on lead in  $\text{PbSO}_4$  is



- \_\_\_\_\_ 7. How many more D atoms should be on the right side of the equation?
- \_\_\_\_\_ 8. What is the formula of the missing product molecule?
- \_\_\_\_\_ 9. What are particles on the left side of the arrow called?
10. Write the completed chemical equation for the reaction pictured.

**Balance the following reactions.**



# Modeling Chemistry Unit 7 Test Answers

**Sebastian Brünink**



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