

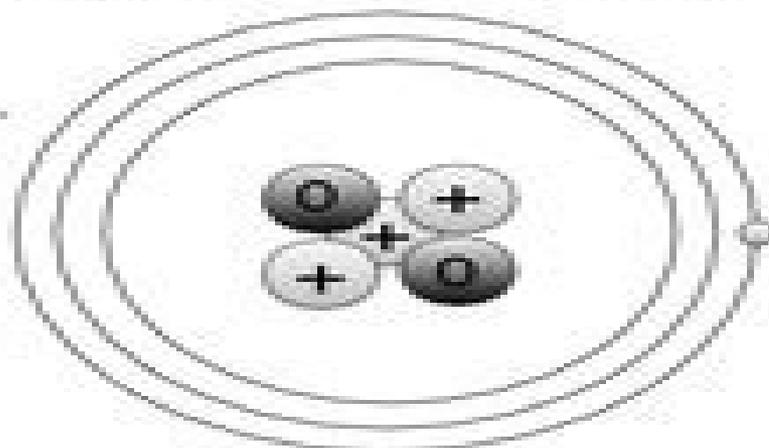
## Structure of an Atom Guided Notes

Atoms are the building blocks of **matter**. They are the smallest particle into which an **element** can be **divided** and still retain its identity. An **element** is matter that is made up of only one type of atom.

Atoms have substructure of their own. They have a **nucleus**, which contains **protons** and **neutrons**. Protons have a **positive charge** and a mass of **1** amu (atomic mass unit). Neutrons have **no charge** and are

also measured at 1 amu. The number of protons in the atomic nucleus, which is the **atomic number**, is the defining characteristic of each element.

**Electrons**, which have a **negative charge** and a mass of almost zero, are located in an area surrounding the nucleus called the **electron cloud**. The way electrons react between charged particles explains the way atoms form **molecules**.



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Matter atomic number electrons molecules no charge positive charge  
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# Guided Notes The Atom

**Sabine Zange**



## Guided Notes The Atom:

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- Chapter 4: Guided Notes The Atom in Specific Contexts
- Chapter 5: Conclusion

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5. In chapter 4, the author will scrutinize the relevance of Guided Notes The Atom in specific contexts. This chapter will explore how Guided Notes The Atom is applied in specialized fields, such as education, business, and technology.
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