

**CHAPTER 6 REVIEW***Chemical Bonding***MIXED REVIEW**

**SHORT ANSWER** Answer the following questions in the space provided.

1. Name the type of energy that is a measure of strength for each of the following types of bonds:

\_\_\_\_\_ **lattice energy** \_\_\_\_\_ a. ionic bond  
 \_\_\_\_\_ **bond energy** \_\_\_\_\_ b. covalent bond  
 \_\_\_\_\_ **enthalpy of vaporization** \_\_\_\_\_ c. metallic bond

2. Use the electronegativity values shown in **Figure 20**, on page 161 of the text, to determine whether each of the following bonds is nonpolar covalent, polar covalent, or ionic.

\_\_\_\_\_ **ionic** \_\_\_\_\_ a. H—F \_\_\_\_\_ **nonpolar covalent** \_\_\_\_\_ d. H—H  
 \_\_\_\_\_ **ionic** \_\_\_\_\_ b. Na—Cl \_\_\_\_\_ **polar covalent** \_\_\_\_\_ e. H—C  
 \_\_\_\_\_ **polar covalent** \_\_\_\_\_ c. H—O \_\_\_\_\_ **polar covalent** \_\_\_\_\_ f. H—N

3. How is a hydrogen bond different from an ionic or covalent bond?

A hydrogen bond is a dipole-dipole attraction between a partially positive hydrogen atom and the unshared electron pair of a strongly electronegative atom such as O, N, or F. Unlike ionic or covalent bonds, in which electrons are given up or shared, the hydrogen bond is a weaker attraction. Hydrogen bonds are generally intermolecular, while ionic and covalent bonds occur between ions or atoms respectively.

4. H<sub>2</sub>S and H<sub>2</sub>O have similar structures and their central atoms belong to the same group. Yet H<sub>2</sub>S is a gas at room temperature and H<sub>2</sub>O is a liquid. Use bonding principles to explain why this is.

Oxygen has higher electronegativity than sulfur, which creates a highly polar bond. Increased polarity in H<sub>2</sub>O bonds means a stronger intermolecular attraction, making water a liquid at room temperature. Hydrogen bonding exists between water molecules, but not between hydrogen sulfide molecules.

# Chapter 8 Covalent Bonding Test B Answers

**Peipei Pang**



**Chapter 8 Covalent Bonding Test B Answers:**

## Unveiling the Magic of Words: A Report on "**Chapter 8 Covalent Bonding Test B Answers**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Chapter 8 Covalent Bonding Test B Answers**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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### **Table of Contents Chapter 8 Covalent Bonding Test B Answers**

1. Understanding the eBook Chapter 8 Covalent Bonding Test B Answers
  - The Rise of Digital Reading Chapter 8 Covalent Bonding Test B Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 8 Covalent Bonding Test B Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 8 Covalent Bonding Test B Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 8 Covalent Bonding Test B Answers
  - Personalized Recommendations
  - Chapter 8 Covalent Bonding Test B Answers User Reviews and Ratings

- Chapter 8 Covalent Bonding Test B Answers and Bestseller Lists
- 5. Accessing Chapter 8 Covalent Bonding Test B Answers Free and Paid eBooks
  - Chapter 8 Covalent Bonding Test B Answers Public Domain eBooks
  - Chapter 8 Covalent Bonding Test B Answers eBook Subscription Services
  - Chapter 8 Covalent Bonding Test B Answers Budget-Friendly Options
- 6. Navigating Chapter 8 Covalent Bonding Test B Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Chapter 8 Covalent Bonding Test B Answers Compatibility with Devices
  - Chapter 8 Covalent Bonding Test B Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chapter 8 Covalent Bonding Test B Answers
  - Highlighting and Note-Taking Chapter 8 Covalent Bonding Test B Answers
  - Interactive Elements Chapter 8 Covalent Bonding Test B Answers
- 8. Staying Engaged with Chapter 8 Covalent Bonding Test B Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chapter 8 Covalent Bonding Test B Answers
- 9. Balancing eBooks and Physical Books Chapter 8 Covalent Bonding Test B Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chapter 8 Covalent Bonding Test B Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 8 Covalent Bonding Test B Answers
  - Setting Reading Goals Chapter 8 Covalent Bonding Test B Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 8 Covalent Bonding Test B Answers
  - Fact-Checking eBook Content of Chapter 8 Covalent Bonding Test B Answers
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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