

## 9

## CHEMICAL NAMES AND FORMULAS

**SECTION 9.1 NAMING IONS (pages 253–258)**

*This section explains the use of the periodic table to determine the charge of an ion. It also defines polyatomic ion and gives the names and formulas for the most common polyatomic ions.*

**► Monatomic Ions (pages 253–256)**

1. What are monatomic ions?

*Monatomic ions are ions consisting of only one atom.*

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2. How is the ionic charge of a Group 1A, 2A, or 3A ion determined?

*The ionic charge is numerically equal to the group number.*

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3. How is the ionic charge of a Group 5A, 6A, or 7A ion determined?

*The charge of an ion in Groups 5A, 6A, or 7A is determined by subtracting 8 from the group number.*

---

4. Circle the letter of the type of element that often has more than one common ionic charge.

- a. alkali metal
- b. alkaline earth metal
- c. transition metal
- d. nonmetal

5. The \_\_\_\_\_ **Stock system** \_\_\_\_\_ of naming transition metal cations uses a Roman numeral in parentheses to indicate the numeric value of the ionic charge.

6. An older naming system uses the suffix **-ous** to name the cation with the \_\_\_\_\_ **lesser** \_\_\_\_\_ charge, and the suffix **-ic** to name the cation with the \_\_\_\_\_ **greater** \_\_\_\_\_ charge.

7. What is a major advantage of the Stock system over the old naming system?

*The Stock system gives the actual charge of the ion.*

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# Chemistry Chapter 9 Chemical Names Formulas Answers

**Liying Dong**



## **Chemistry Chapter 9 Chemical Names Formulas Answers:**

*American Druggist*, 1927      *Foundations of College Chemistry* Morris Hein, Susan Arena, 2010-01-26 Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years this book has helped them master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis      *Chemistry for Boys and Girls* Carlotta Cherryholmes Greer, Julia Cora Bennett, 1925      **Chemistry** John A. Olmsted, Gregory M. Williams, Robert Charles Burk, 2016-01-14 Olmsted Burk is an introductory general chemistry text designed specifically with Canadian professors and students in mind A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings It more accurately reflects the curriculum of most Canadian institutions Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant      Understanding Chemistry Fred M. Dewey, 1994      ... Chemistry Jay Ellis Stannard, 1927      **Meyer Brothers Druggist**, 1896      *Pharmaceutical Record and Weekly Market Review* P. W. Bedford, 1886      **Study Guide/Selected Solutions Manual** Julie R. Frentrup, 2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book      **State board examination questions & answers of the United States and Canada** Medical record, 1912      *Druggists' Circular*, 1920      Chemistry Made Clear R. Gallagher, P. Ingram, 1987 Chemistry Made Clear is widely used as a core GCSE Chemistry text or as the Chemistry component of a balanced science course Students will be able to find things out quickly and easily among the simplified explanations Each double page spread deals with a different topic and includes questions Exam level questions at the end of each chapter Line drawings and photographs highlight the real life applications of chemistry      The Pharmaceutical Era, 1916      *American Druggist and Pharmaceutical Record*, 1915      *Rocky Mountain Druggist*, 1903      Professional Examination Papers, 1905      *College Department Bulletins* University of the State of New York, 1909 Includes universities professional and technical schools      *The Northwestern Druggist*, 1905      **The Electrical Experimenter**, 1919      **Pharmacy State Board Questions and Answers, 1963** Ralph J. Mill, 1962

## **Chemistry Chapter 9 Chemical Names Formulas Answers** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Chemistry Chapter 9 Chemical Names Formulas Answers**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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