

CHEMISTRY MIDTERM REVIEW

1.) CONVERSIONS IN METRIC UNITS

How many kilograms are in 143 grams?

Using KHDBDCM

Starting with Grams (g) – base unit. Moving to kilograms (kg); move the decimal three spaces to the right

143 g – 0.143 g

Using conversion factors

KNOW that kilo = 1000 base

$143 \text{ g} \times (1 \text{ kg}/1000\text{g}) = 0.143 \text{ kg}$

2.) CONVERSIONS WITH ENGLISH UNITS

How many milliliters are in 13.5 oz?

MUST USE CONVERSION FACTORS (oz, lb, feet, inches...)

Always start with what you know... Figure out where to go

13.5 oz

? mL

13.5 oz	1 cup	1 pt	1 qt	1 L	1000 mL
	8 oz	2 cups	2 pt	1.06 qt	1 L

= 398 mL are in 13.5 oz

Question: is ATM (atmospheres) a standard number?

Answer; No, it's a unit. There's not a particular number there. The gas constant has a number and a unit because it is a constant; atmospheres will differ depending on the problem.

Stoichiometry: Practice questions.

22.0 g of propane will combine with what mass (g) of O₂?



Chemistry Midterm Review Answers

Jessica J Manson



Chemistry Midterm Review Answers:

Decoding **Chemistry Midterm Review Answers**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Chemistry Midterm Review Answers**," a mesmerizing literary creation penned with a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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