

# STUDY GUIDE

## SECTION 1.1 - 1.4

- **matter**: everything that takes up space + has a mass

### Properties of matter



### ① Physical changes:

- Phase
- Size
- Texture

→ Reversible

### ② Chemical changes:

- Color change
- New substance is formed

→ Irreversible

### Pure substance vs. Mixture

#### MOLECULE (pure substances)

##### Compounds

2 or more  $\neq$  atoms  
(eg:  $H_2O$ ;  $Na_2O$ )

##### Elements

1 atom  
(eg:  $H_2$ ,  $O_2$ )

- **Pure substance**: constant composition + definite properties

# Chemistry Midterm Study Guide

**N Colangelo**



## **Chemistry Midterm Study Guide:**

## Decoding **Chemistry Midterm Study Guide**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Chemistry Midterm Study Guide**," a mesmerizing literary creation penned by a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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