

## Chapter Test II

### A. Matching

Match each item in Column I with the correct description in Column II. Write the letter of the correct description.

#### Column I

- \_\_\_\_\_ 1. An unbalanced chemical equation for a reaction that obeys the law of conservation of matter.
- \_\_\_\_\_ 2. The number of protons in the nucleus of an atom of an element.
- \_\_\_\_\_ 3. The number of electrons in a neutral atom.
- \_\_\_\_\_ 4. The ratio of the actual weight of an element to the standard weight of 16.
- \_\_\_\_\_ 5. The number of protons and neutrons in the nucleus of an atom.
- \_\_\_\_\_ 6. The number of protons and neutrons in the nucleus of an atom.

#### Column II

- a. atomic weight
- b. atomic number
- c. molecular weight
- d. chemical formula
- e. mass number
- f. relative atomic mass

### B. Multiple Choice

Choose the correct answer for each question.

- \_\_\_\_\_ 1. The number of protons in an atom of an element is equal to the number of
  - a. neutrons in the nucleus
  - b. electrons in the atom
  - c. total mass of the atom
  - d. total number of nucleons in the nucleus
- \_\_\_\_\_ 2. Which of the following is not a chemical formula?
  - a.  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
  - b.  $\text{H}_2\text{O}$
  - c.  $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
  - d.  $\text{H}_2\text{O}$
- \_\_\_\_\_ 3. The ratio of the actual weight of an element to the standard weight of 16 is called
  - a. atomic weight
  - b. relative atomic mass
  - c. atomic number
  - d. mass number

# Chemistry Chapter 12 Stoichiometry Test

**Xiang Xie**



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