

## Chemical Bonding

### SECTION 3

**SHORT ANSWER** Answer the following questions in the space provided.

- A** The notation for sodium chloride, NaCl, stands for one  
(a) formula unit. (c) crystal.  
(b) molecule. (d) atom.
- D** In a crystal of an ionic compound, each cation is surrounded by a number of  
(a) molecules. (c) dipoles.  
(b) positive ions. (d) negative ions.
- B** Compared with the neutral atoms involved in the formation of an ionic compound, the crystal lattice that results is  
(a) higher in potential energy. (c) equal in potential energy.  
(b) lower in potential energy. (d) unstable.
- B** The lattice energy of compound A is greater in magnitude than that of compound B. What can be concluded from this fact?  
(a) Compound A is not an ionic compound.  
(b) It will be more difficult to break the bonds in compound A than those in compound B.  
(c) Compound B has larger crystals than compound A.  
(d) Compound A has larger crystals than compound B.
- B** The forces of attraction between molecules in a molecular compound are generally  
(a) stronger than the attractive forces among formula units in ionic bonding.  
(b) weaker than the attractive forces among formula units in ionic bonding.  
(c) approximately equal to the attractive forces among formula units in ionic bonding.  
(d) equal to zero.
- Describe the force that holds two ions together in an ionic bond.

**Very strong overall attraction between positive and negative charges**

- What type of energy best represents the strength of an ionic bond?

**Lattice energy**

- What types of bonds are present in an ionic compound that contains a polyatomic ion?

# Chemistry Chapter 6 Review Answers

**David Baud**



**Chemistry Chapter 6 Review Answers:**

## Embracing the Beat of Term: An Psychological Symphony within **Chemistry Chapter 6 Review Answers**

In a world used by displays and the ceaseless chatter of instantaneous conversation, the melodic splendor and mental symphony created by the written word frequently fade in to the background, eclipsed by the constant sound and interruptions that permeate our lives. But, situated within the pages of **Chemistry Chapter 6 Review Answers** an enchanting literary prize brimming with organic emotions, lies an immersive symphony waiting to be embraced. Crafted by an outstanding musician of language, that captivating masterpiece conducts visitors on a psychological trip, skillfully unraveling the hidden songs and profound affect resonating within each carefully constructed phrase. Within the depths of the poignant analysis, we will discover the book is central harmonies, analyze their enthralling publishing model, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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