

# COVALENT BONDING

Name \_\_\_\_\_

Covalent bonding occurs when two or more nonmetals share electrons, attempting to attain a stable octet of electrons at least part of the time. For example:

*H wants 2 electrons  
all other atoms 8*



*electrons in the middle share*

Show how covalent bonding occurs in each of the following pairs of atoms. Atoms may share one, two or three pairs of electrons.

*wanted available share*

<p>1. H + H (H<sub>2</sub>)</p> <p>W - 4</p> <p>A - 2</p> <p>S = 2</p>	<p><i>wanted 2 + 2</i></p> <p><i>available</i></p> <p><i>share</i></p> <p><math>\text{H} \cdot \cdot \text{H}</math></p> <p><math>\text{H} \text{---} \text{H}</math></p>	<p><math>\text{H}-\text{H}</math></p> <p>single bond = share 2 e<sup>-</sup></p>
<p>2. F + F (F<sub>2</sub>)</p> <p>W - 8 + 8 = 16</p> <p>A - 7 + 7 = 14</p> <p>S = 2</p>	<p><math>:\ddot{\text{F}}:\ddot{\text{F}}:</math></p> <p><math>\text{F}-\text{F}</math></p>	<p>single bond 2e<sup>-</sup></p>
<p>3. O + O (O<sub>2</sub>)</p> <p>W - 8 + 8 = 16</p> <p>A - 6 + 6 = 12</p> <p>S = 4</p>	<p><math>:\ddot{\text{O}}::\ddot{\text{O}}:</math></p> <p><math>\text{O}=\text{O}</math></p>	<p>4 double bond</p>
<p>4. N + N (N<sub>2</sub>)</p> <p>W - 8 + 8 = 16</p> <p>A - 5 + 5 = 10</p> <p>S = 6</p>	<p><math>:\text{N}:::\text{N}:</math></p> <p><math>\text{N}\equiv\text{N}</math></p>	
<p>5. C + O (CO<sub>2</sub>)</p> <p>W - 8 + 8 + 8 = 24</p> <p>A - 4 + 6 + 6 = 16</p> <p>S = 8</p>	<p><math>:\ddot{\text{O}}:::\text{C}:::\ddot{\text{O}}:</math></p> <p><math>\ddot{\text{O}}=\text{C}=\ddot{\text{O}}</math></p>	
<p>6. H + O (H<sub>2</sub>O)</p> <p>W - 2 + 2 + 8 = 12</p> <p>A - 1 + 1 + 6 = 8</p> <p>S = 4</p>	<p><math>\text{H}:\ddot{\text{O}}:\text{H}</math></p> <p><math>\text{H}-\ddot{\text{O}}-\text{H}</math></p>	

# Chemistry Covalent Bonding Packet Answers

**Mike Jess**



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