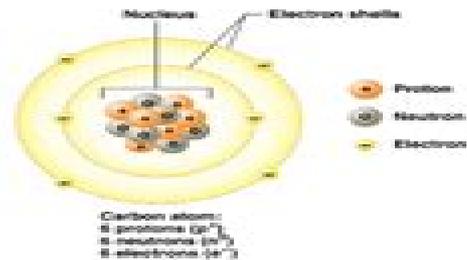


The Chemistry of Life

Atoms and atomic structure

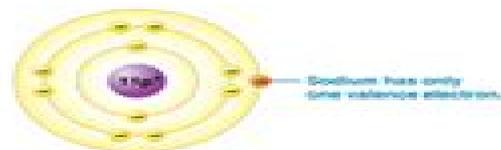
- **Atoms:** smallest, stable unit of matter
 - Subatomic particles make up atoms
- **Atomic nucleus:**
 - **Protons (p^+):** Positive electrical charge
 - **Neutrons (n or n^0):** No electrical charge (neutral)
- **Electron cloud (shell) outside nucleus:**
 - **Electrons (e^-):** Negative electrical charge. Much smaller than protons or neutrons.



- **Atoms are electrically neutral:** The number of positively charged protons in an atom equals the number of negatively charged electrons.

Electron shells (clouds): Electrons circle nucleus. Circle in orbits (shells) numbered 1-7

- 1st shell: closest to nucleus, can hold 2 electrons
- 2nd shell: can hold 8 electrons
- 3rd shell: can hold 18 electrons but "satisfied" with 8



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JE Gale



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