

Grade 11 Chemistry Exam Review

1. Fill in the following chart with the number of electrons, protons, neutrons, atomic number, mass number, and the element notations

Atom	Atomic Number	Mass Number	# of protons	# of neutrons	# of electrons
K ⁺					
Es					
Ms					
P ³⁻	13		13	14	
		75		42	
		184	74	117	78
	99	252			
		280		157	

2. Draw Bohr diagrams and Lewis dot diagrams for the following elements: Li, Ca, Al, C, P, O, F, K
 3. Draw Bohr diagrams of the following ions: Na⁺, Mg²⁺, B³⁺, Sr²⁺, N³⁻, S²⁻, Cl⁻
 4. Find the average atomic mass of P when the atomic mass with 15 neutrons is 30 is in 77% abundance and the atomic mass with 17 neutrons is 32 is in 23% abundance. (30.46 amu)
 5. Find the Average atomic mass of Ca when the element which has 20 neutrons is in 12% abundance and the element which has 21 neutrons is in 88% abundance. (40.88 amu)
 6. Find the average atomic mass of Cr if the atomic number is 24 and the two isotopes are; 18 neutrons at 97% abundance, and 19 neutrons at 3% abundance. (42.03 amu)
 7. What is ionization energy, electron affinity, and electronegativity?
 8. For the following elements decide which one has a larger atomic radius, electronegativity, ionization energy, and reactivity.

		Atomic Radius largest	electronegativity largest	ionization energy largest	reactivity greatest
O	Po				
B	N				
Ni	Pt				
Mg	Ra				
K	Fe				
Ag	Rb				
Au	Cu				
Hg	Zn				
I	Cl				
As	N				

9. Describe how the atomic radius gets smaller as we add more electrons in a row.
 10. Describe how the reactivity is linked to ionization energy for the metals and electron affinity for the non-metals.
 11. Which family of elements do the following elements belong?
 a) He _____
 c) Ca _____
 e) Pm _____
 g) Yb _____
 i) Np _____
 k) Fr _____
 b) Cl _____
 d) Cs _____
 f) Es _____
 h) I _____
 j) Sr _____
 l) Xe _____
 12. What is the octet rule and what does it mean for any element that has its octet full.
 13. What is the bonding capacity of the following elements: P, S, Cl, Na, K, Ba, B, Mg, Li, Si
 14. Define the following terms: alloy, metalloid, solution, hydrate
 15. How is the matter classified?
 16. Describe the difference between ionic bonds and covalent bonds. Be sure to include which type of elements are joined together for each type of bond, how strong the bond is, if the electrons are either transferred or shared.
 17. Indicate whether the following are ionic or covalent bonds and draw the lewis structures would look with the bonds.
 a) LiF _____
 e) AlF₃ _____
 b) F₂ _____
 f) CO₂ _____
 c) MgS _____
 g) H₂O _____
 d) O₂ _____
 h) Mg₃N₂ _____

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Guillaume Favre



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Ultimate Guide for FCI Assistant Grade - III Recruitment

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2 The Data Interpretation section has been provided for paper 2 *Oswaal CDS (Combined Defence Services) 14 Solved Papers Year-wise 2018-2024 (II) | General Knowledge | For 2025 Exam* Oswaal Editorial Board,2024-09-26 Union Public Service Commission UPSC every year conducts a CDS exam twice a year for candidates who wish to make their career in the defence forces Army Navy and Air Force The Combined Defence Services Examination is conducted for admission to the Indian Military Academy IMA Indian Naval Academy INA Air Force Academy AFA and Officers Training Academy OTA The CDS selection process comprises two stages written exams and SSB interviews The final selection of candidates is done based on the performance in both stages After completing training at IMA INA AFA and OTA candidates are selected for the post of Lieutenant In 2024 Approx 4 5 Lacs students applied for the CDS examination the opportunity you get from the Indian Armed Forces is just limitless which helps in enhancing your personality traits For a youngster who is aspiring to get a job full of challenges and excitement then there is no better job than the defence This book aims to make aspirants exam ready boost their confidence and help them achieve better results in CDS By making learning Simple we are also making better careers and a better life for every student Every day we are moving ahead pursuing our noble cause of spreading knowledge This set of solved question papers is designed to enrich students with ample and examoriented practice so that they can clear CDS Examination with extraordinary results Not one or two but 14 Previous Year Solved Question Paper 2018 to 2024 II focussed on polishing every topic Thorough studying of this book will boost my confidence and familiarise me with exam patterns Some benefits of studying from Oswaal CDS check 14 Previous year solved question papers 100% updated with Fully Solved Paper of September 2024 II Concept Clarity with detailed explanations of 2018 to 2024 I Papers Extensive Practice with 1600 Questions and Two Sample Question Papers Crisp Revision with Mind Maps Expert Tips helps you get expert knowledge master crack CDS in first attempt Exam insights with Previous Years 2024 2019 Trend Analysis empowering students to be 100% exam ready Our Heartfelt Gratitude Finally we would like to thank our authors editors and reviewers Special thanks to our students who send us suggestions and constantly help improve our books To stay true to our motto of Learning Made Simple we constantly strive to present information in ways that are easy to understand as well as remember [Oswaal CDS \(Combined Defence Services\) 14 Solved Papers Year-wise 2018-2024 \(II\) | English | For 2025 Exam](#) Oswaal Editorial Board,2024-09-26 Union Public Service Commission UPSC every year conducts a CDS exam twice a year for candidates who wish to make their career in the defence forces Army Navy and Air Force The Combined Defence

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