

**Chemistry**  
**Chapter 9 Practice Test: Stoichiometry**

**Name** \_\_\_\_\_

**Free Response**

*Use the following equation to answer problems 1 - 4 at STP. (Show All Work)*



1. How many **moles** of aluminum chloride ( $\text{AlCl}_3$ ) would be produced from 0.12 moles of Hydrochloric acid (HCl)?
2. If you wanted to completely react 5.0 moles of Aluminum, how many **moles** of hydrochloric acid (HCl) would be needed?
3. How many **grams** of hydrochloric acid (HCl) would be needed to completely react 35.8 g of Aluminum foil?
4. How many **grams** of aluminum would be needed to produce 67.2 dm<sup>3</sup> of hydrogen gas?

*Use the following equation to answer problems 5 and 6 at STP.*



5. How many **molecules** of carbon dioxide ( $\text{CO}_2$ ) would be produced by completely reacting 8.00 grams of oxygen gas ( $\text{O}_2$ ) with excess Carbon Monoxide (CO)?
6. What **volume** in L of oxygen gas ( $\text{O}_2$ ) would react with 6.0 L of carbon monoxide gas (CO) ?

# Chemistry Chapter 9 Stoichiometry Test

**Henry Augustus Mott**



## **Chemistry Chapter 9 Stoichiometry Test:**

**McGraw-Hill's SAT Subject Test Chemistry, 3rd Edition** Thomas Evangelist, 2011-02-03 Expert guidance on the Chemistry exam Many colleges and universities require you to take one or more SAT II Subject Tests to demonstrate your mastery of specific high school subjects McGraw Hill's SAT Subject Test Chemistry is written by experts in the field and gives you the guidance you need to perform at your best This book includes 4 full length sample tests updated for the latest test format 40 top tips to remember on test day Glossary of tested chemistry terms and formulas Tips and strategies from one of the most popular teachers at the renowned Brooklyn Technical High School Diagnostic test to pinpoint strengths and weaknesses Step by step review of all topics covered on the exam In depth coverage of the lab experiment questions that are a major test feature Charts tables and illustrations to simplify and reinforce learning Practice tests just like the real SAT Subject Test in Chemistry Test taking tips and strategies

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