

CBSE Sample Paper Class 12 Chemistry Set 1

Time allowed: 3 Hours

Max. Marks: 70

General Instructions:

- (a) All questions are compulsory
- (b) Section A: Q. no. 1 to 5 are very short questions and carry 1 mark each.
- (c) Section B: Q. no. 6 to 12 are short answer questions and carry 2 marks each.
- (d) Section C: Q. no. 13 to 24 are also short answer questions and carry 3 marks each.
- (e) Section D: Q. no. 25 to 27 are long answer questions and carry 5 marks each.
- (f) There is no overall choice. However an internal choice has been provided in two questions of one mark, two questions of two marks, four questions of three marks and all the three questions of five marks weightage. You have to attempt only one of the choices in such questions.
- (g) Use of log tables if necessary, use of calculators is not allowed.

Section-A (1 mark each)

Q1. What is the Normality of 1.5 M H_2SO_4 ?

OR

Why Al cannot be obtained by electrolysis of fused AlCl_3 ?

Q2. A reaction is found to be zero order. Will its molecularity be also zero.

Q3. Which out of Molarity or Molality varies with temperature?

OR

Define Adsorption.

Q4. Which method is used to concentrate the sulphide ores?

Q5. Name the elements of the 13th group and give general electronic configuration of their valence shells.

Section-B (2 marks each)

Q6. In a close packing of N spheres, how many (i) tetrahedral and (ii) octahedral sites are present?

OR

What is the role of salt-bridge in an electrochemical cell?

Q7. Mention any four factors affecting rate of a reaction.

OR

Describe the two uses of each of the following:

- (a) Copper sulphate (b) Silver nitrate
 - (i) For copper plating
 - (ii) As germicide and fungicide
- (b) Uses of Silver nitrate

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Victor M. Corman



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Table of Contents Chemistry Question Paper For Class 12 Cbse 2012

1. Understanding the eBook Chemistry Question Paper For Class 12 Cbse 2012
 - The Rise of Digital Reading Chemistry Question Paper For Class 12 Cbse 2012
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Question Paper For Class 12 Cbse 2012
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Question Paper For Class 12 Cbse 2012
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Question Paper For Class 12 Cbse 2012
 - Personalized Recommendations

- Chemistry Question Paper For Class 12 Cbse 2012 User Reviews and Ratings
- Chemistry Question Paper For Class 12 Cbse 2012 and Bestseller Lists
- 5. Accessing Chemistry Question Paper For Class 12 Cbse 2012 Free and Paid eBooks
 - Chemistry Question Paper For Class 12 Cbse 2012 Public Domain eBooks
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 - Chemistry Question Paper For Class 12 Cbse 2012 Budget-Friendly Options
- 6. Navigating Chemistry Question Paper For Class 12 Cbse 2012 eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Question Paper For Class 12 Cbse 2012 Compatibility with Devices
 - Chemistry Question Paper For Class 12 Cbse 2012 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Question Paper For Class 12 Cbse 2012
 - Highlighting and Note-Taking Chemistry Question Paper For Class 12 Cbse 2012
 - Interactive Elements Chemistry Question Paper For Class 12 Cbse 2012
- 8. Staying Engaged with Chemistry Question Paper For Class 12 Cbse 2012
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Question Paper For Class 12 Cbse 2012
- 9. Balancing eBooks and Physical Books Chemistry Question Paper For Class 12 Cbse 2012
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Question Paper For Class 12 Cbse 2012
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Question Paper For Class 12 Cbse 2012
 - Setting Reading Goals Chemistry Question Paper For Class 12 Cbse 2012
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Question Paper For Class 12 Cbse 2012
 - Fact-Checking eBook Content of Chemistry Question Paper For Class 12 Cbse 2012

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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